

## ENANTIOSELECTIVE SYNTHESSES OF (-)-SEMBURIN AND (-)-ISOSEMBURIN

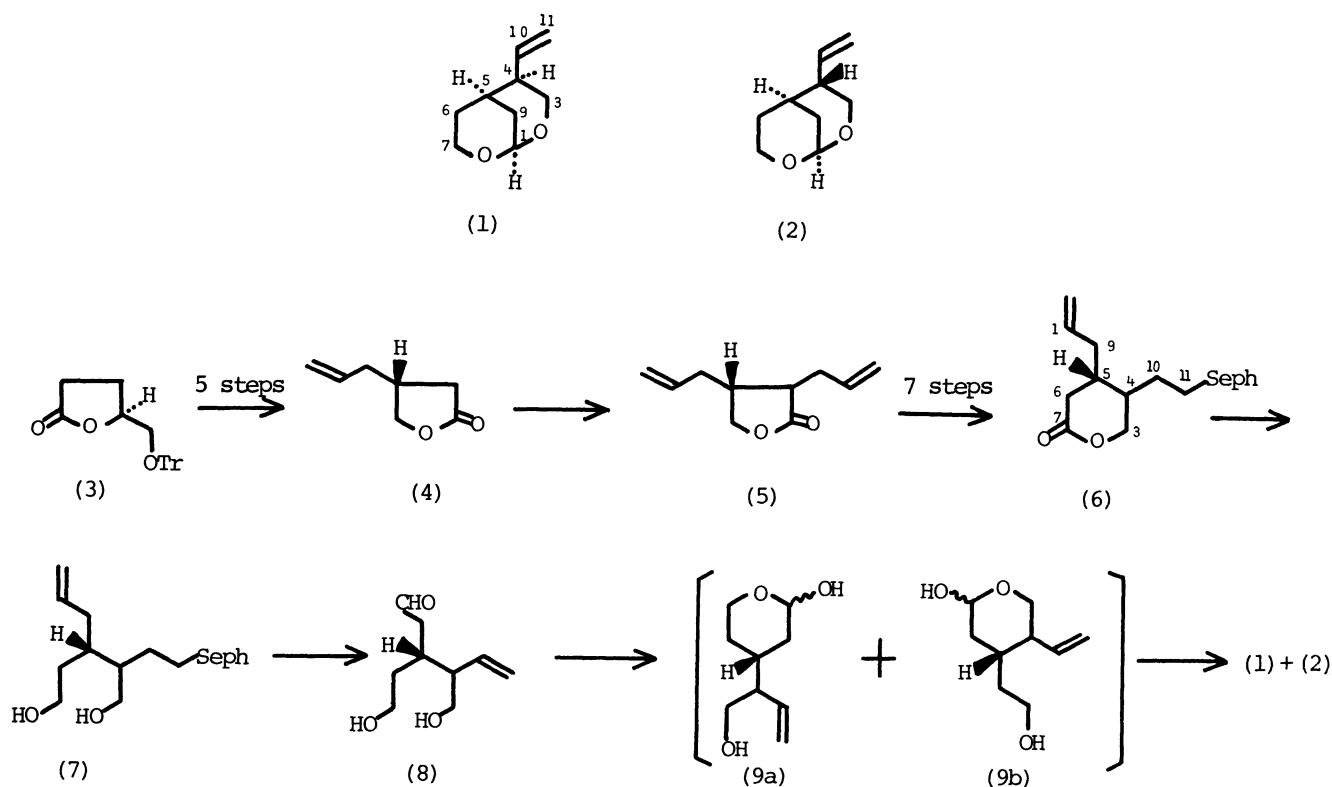
Seichi TAKANO<sup>\*</sup>, Nobuhiko TAMURA, Kunio OGASAWARA,  
Yoshiko NAKAGAWA<sup>†</sup>, and Tsutomu SAKAI<sup>†</sup>

Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980  
and <sup>†</sup>Suntory Institute for Bioorganic Research, Shimamoto-cho,  
Mishima-gun, Osaka 618

Validity of the proposed structures of (-)-semburin(1) and (-)-isosemburin(2), isolated from Swertia japonica Makino, has been established by the enantioselective total syntheses starting from the known chiral lactone(3).

The structures of (-)-semburin and (-)-isosemburin, both the constituents of the volatile oil of Swertia japonica Makino, were proposed to be unique 2,8-dioxabicyclo[3.3.1]nonane frameworks (1) and (2), respectively, based on NMR spectroscopy and biogenetic considerations.<sup>1)-3)</sup> We now report here the enantioselective syntheses of both (-)-semburin(1) and (-)-isosemburin(2), starting from the chiral lactone(3)<sup>4)</sup> with known absolute configuration, which could unambiguously confirm the appropriateness of the proposed structures.

According to the established thirteen step procedure,<sup>5)</sup> the  $\gamma$ -lactone(3) was converted into the  $\delta$ -lactone(6:4( $\alpha$ )H) along with a minor amount of its 4( $\beta$ )H-epimer (6:4( $\beta$ )H) owing to imperfect stereoselectivity in the second alkylation stage((4) $\rightarrow$ (5)). The obtained epimeric mixture, without separation, was reduced with diisobutylaluminum hydride(THF, -78°C - room temperature), followed by sodium borohydride to give the diol(7)<sup>6)</sup> as an epimeric mixture. The epimeric diol(7) was ozonized(MeOH, -78°C), followed by treating with an excess of triethylamine<sup>7)</sup>(-78°C - room temperature) to yield the vinyl-aldehyde(8) which formed a mixture of the hemiacetals, (9a) and (9b), spontaneously during the work-up. The mixture, upon reflux azeotropically in benzene in the presence of a catalytic amount of *p*-toluenesulfonic acid, yielded a mixture of (-)-semburin(1) and (-)-isosemburin(2) as an oil possessing camphor like odor. Separation of the mixture using a preparative gas chromatography(10% Thermon-1500 on Chromosorb W(AW-DMCS)) gave (-)-semburin(1),  $[\alpha]_D^{23.5} -1.89^\circ$ ( $c=0.53$ , CHCl<sub>3</sub>)(natural,  $[\alpha]_D^{25.5} -2.00^\circ$ ( $c=0.10$ ,



### Scheme

$\text{CHCl}_3$ )), as a major component and (-)-isosemburin(2),  $[\alpha]_{\text{D}}^{23.5} -9.71^\circ (c=0.35, \text{CHCl}_3)$  (natural,  $[\alpha]_{\text{D}}^{25} -8.0^\circ (c=0.05, \text{CHCl}_3)$ ), as a minor component. Their GC and spectroscopic (IR,  $^1\text{H-NMR}$  (360MHz), MS) properties were all identical with those of the corresponding natural products.

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### References

1. T. Sakai, H. Naoki, K. Takaki, and H. Kameoka, *Chemistry Lett.*, **1981**, 1257.
2. The reported optical rotation of semburin(1),  $[\alpha]_{\text{D}}^{24} +6.0^\circ (c=0.05, \text{CHCl}_3)$  (see, ref. 1), must be corrected and it should read  $[\alpha]_{\text{D}}^{25.5} -2.00^\circ (c=0.10, \text{CHCl}_3)$ .
3. Synthesis of racemic semburin(1) has been recently accomplished by employing the fundamentally different approach: Y. Ohfune, K. Takaki, H. Kameoka, and T. Sakai, *Chemistry Lett.*, **1982**, 209.
4. (a) K. Tomioka and K. Koga, *Tetrahedron Lett.*, 3315(1979).  
(b) S. Takano, M. Yonaga, and K. Ogasawara, *Synthesis*, 265(1981) and references cited therein.
5. S. Takano, N. Tamura, and K. Ogasawara, *J. Chem. Soc., Chem. Comm.*, 1155(1981).
6. Reduction using lithium aluminum hydride alone in THF also gave the diol(7) accompanied by a considerable amount of deselenated by-products.
7. H. Iio, M. Isobe, T. Kawai, and T. Goto, *Tetrahedron*, **35**, 941(1979).

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